
Biochemistry Explained

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haemoglobin a 1c, hba1c, glycated haemoglobin, glycohaemoglobin, ghb. 1.3 nlm code to follow 1.4 description of analyte hba1c is a modified haemoglobin, with a stable adduct of glucose (covalently ... **lecture 3: enzyme kinetics - inf.ed** - of biochemistry, iv edition, w. h. freeman ed. computational systems biology 22 ph dependence for chymotrypsin • chymotrypsin (another enzyme that hydrolyzes peptide bonds) has an optimum of activity at ph 8 • the activity alteration is linked to the ionization state of the aminoacid side chains in the active site. • the transitions in (b) at ph 7.5 and (c) at ph 8.5 are due to ... **routine blood tests: part 1 of 4 why do we test for urea ...** - commonly requested biochemistry tests. they provide essential information on renal function, principally in excretion and homeostasis. creatinine levels are a major factor in determining the estimated glomerular filtration rate, which is the gold standard marker of kidney health. the tests' clinical value is in the diagnosis and management of acute kidney injury and chronic kidney disease ... **an idea to explore: understanding redox reactions in ...** - understanding redox reactions in biochemistry requires a clear grasp of three definitions: electron exchange, oxidation number, and oxygen atom insertion. here, the electrochemical cell is explained for redox-active ions, introducing a new comparison between the measurement of midpoint potential and the measurement of initial velocity for enzymes. oxygen atom insertion is shown to be of use ...

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